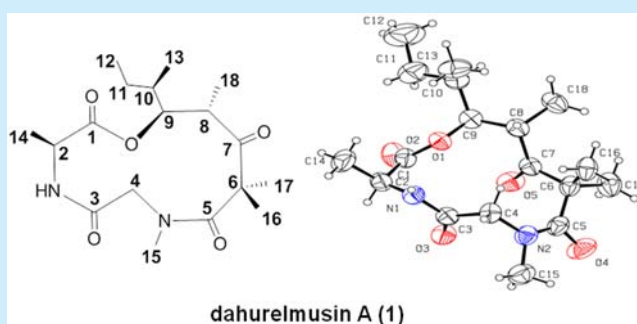


Dahurelmusin A, a Hybrid Peptide–Polyketide from *Elymus dahuricus* Infected by the *Epichloë bromicola* EndophyteQiu-Yan Song,<sup>†</sup> Hai-Tao Yu,<sup>‡</sup> Xing-Xu Zhang,<sup>†</sup> Zhi-Biao Nan,<sup>\*,†</sup> and Kun Gao<sup>\*,§</sup><sup>†</sup>State Key Laboratory of Grassland Agro-Ecosystems, College of Pastoral Agriculture Science and Technology, Lanzhou University, Lanzhou 730020, China<sup>‡</sup>Institute of Plant Protection, Gansu Academy of Agricultural Sciences, Lanzhou 730070, China<sup>§</sup>State Key Laboratory of Applied Organic Chemistry, College of Chemistry and Chemical Engineering, Lanzhou University, Lanzhou 730000, China

## S Supporting Information

**ABSTRACT:** One novel hybrid peptide–polyketide, dahurelmusin A (**1**), was isolated from *Elymus dahuricus* infected by the *Epichloë bromicola* endophyte. Comprehensive spectroscopic analysis revealed that **1** is the first example of hybrid peptide–polyketide possessing an unprecedented 5-hydroxy-2,2,4,6-tetramethyl-3-oxooctanoic acid moiety. The single-crystal X-ray diffraction analyses allowed the absolute configuration assignment of this compound. Compound **1** also exhibited significant insecticidal activities against *Rhopalosiphum padi* and *Brevicoryne brassicae* with LC<sub>50</sub> values of 0.092 and 0.251 mM, respectively.



As one of the cool season grasses, *Elymus dahuricus* (*E. dahuricus*) is widely planted in subtropical and temperate regions.<sup>1</sup> The stems and leaves of *E. dahuricus* are abundant natural resources in China and are used for high-quality herbage purposes, indicating that the grass may have low toxicity to dairy animals and be environmentally biodegradable. The frequency of infection of *E. dahuricus* by *Epichloë* spp. ranges from 20 to 100% in China.<sup>2–5</sup> However, until now, the metabolites of *Epichloë bromicola* (*E. bromicola*) infected *E. dahuricus* and its insecticidal activities have not been investigated. As part of a program with the goal of screening of *Epichloë* spp. infected forage for insecticides, we report herein the isolation and identification of metabolites from an extract of *E. bromicola*-infected *E. dahuricus*. The isolated metabolites were further evaluated for their insecticidal activities against *Rhopalosiphum padi* (*R. padi*) and *Brevicoryne brassicae* (*B. brassicae*).

Compound **1** was obtained as an optically active colorless crystal (Figure 1). The <sup>13</sup>C NMR and DEPT spectra, together

with the molecular ion peak in the HRESIMS, indicated a molecular formula of C<sub>18</sub>H<sub>30</sub>O<sub>5</sub>N<sub>2</sub>, requiring five degrees of unsaturation. The IR spectrum of compound **1** showed absorption bands of carbonyl groups (1719, 1701, and 1687 cm<sup>−1</sup>). The <sup>1</sup>H NMR and DEPT spectra of metabolite **1** revealed the presence of an N–H group at δ<sub>H</sub> 6.89 (d, *J* = 10.0 Hz); an oxygenated methine proton at δ<sub>H</sub> 5.19 (d, *J* = 9.2 Hz); two methine protons at δ<sub>H</sub> 3.67 (m) and 4.60 (m); a set of methylene protons at δ<sub>H</sub> 2.92 and 4.96 (d, *J* = 13.2 Hz); another set of methylene protons at δ<sub>H</sub> 1.06 and 1.41 (m); an N-methyl group at δ<sub>H</sub> 3.16; two tertiary methyl signals at δ<sub>H</sub> 1.62 and 1.76; three secondary methyl groups at δ<sub>H</sub> 0.91 (d, *J* = 6.8 Hz), 1.02 (d, *J* = 7.2 Hz), and 1.32 (d, *J* = 7.2 Hz); and a triplet methyl group at δ<sub>H</sub> 0.94 (t, *J* = 7.2 Hz). The <sup>13</sup>C NMR and DEPT spectra of compound **1** resolved 18 carbon signals, corresponding to four carbonyl carbons (one ketone carbon), one quaternary carbon, four methine carbons (one oxygenated carbon), two methylene carbons, and seven methyl carbons (Table 1). Accordingly, a ring was required for **1** to fulfill the unsaturation requirement.

The presence of characteristic signals for two amino carbons (δ<sub>C/H</sub> 54.9/2.92, 4.96; 49.9/4.60), one N–H group (δ<sub>H</sub> 6.89), one N-methyl group (δ<sub>C/H</sub> 36.5/3.16), and two carbonyl/amide carbonyl carbons (169.8, 170.0) in the NMR spectra of **1** was indicative of a peptide. The <sup>1</sup>H–<sup>1</sup>H COSY spectrum of metabolite **1** revealed a spin-coupling system of the H<sub>3</sub>–14, H-2, and N–H proton, and the HMBC correlation of H<sub>3</sub>–14 (δ<sub>H</sub>

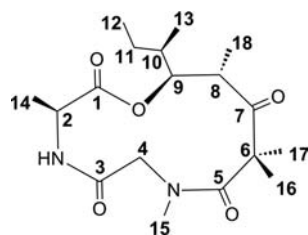


Figure 1. Structure of dahurelmusin A (**1**).

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Table 1.  $^1\text{H}$  and  $^{13}\text{C}$  NMR Data of **1**<sup>a</sup>

	<b>1</b>	
	$\delta_{\text{H}}$ mult (J in Hz)	$\delta_{\text{C}}$
1		169.5
2	4.60 m	49.9
3		169.8
4	2.92 d (13.2) 4.96 d (13.2)	54.9
5		170.0
6		58.3
7		210.0
8	3.67 m	42.1
9	5.19 d (9.2)	77.8
10	1.64 m	36.6
11	1.06 m 1.41 m	26.9
12	0.94 t (7.2)	12.7
13	0.91 d (6.8)	30.6
14	1.32 d (7.2)	21.2
15	3.16 s	36.5
16	1.62 s	22.8
17	1.76 s	25.8
18	1.02 d (7.2)	15.5
NH	6.89 d (10.0)	

<sup>a</sup> $^1\text{H}$  NMR (400 MHz) and  $^{13}\text{C}$  NMR (100 MHz), TMS, measured in  $\text{CDCl}_3$ .

1.32) with C-1 led to the construction of the alanine. Furthermore, the HMBC correlations of H<sub>2</sub>-4 ( $\delta_{\text{H}}$  2.92, 4.96) with C-3 and N-methyl ( $\delta_{\text{H}}$  3.16) with C-4 and C-5 suggested that metabolite **1** also contained an N-methylglycine fragment. In addition, the  $^1\text{H}$ – $^1\text{H}$  COSY spectrum of metabolite **1** revealed a spin-coupling system of H<sub>3</sub>-12, H<sub>2</sub>-11, H-10, H-9, H-8, and H<sub>3</sub>-18, another spin-coupling system of H-10 and H<sub>3</sub>-13; the HMBC correlations of H<sub>3</sub>-16/17 ( $\delta_{\text{H}}$  1.62, 1.76) with C-5, C-6, C-7, and H<sub>3</sub>-18 ( $\delta_{\text{H}}$  1.02) with C-7 indicated that metabolite **1** had a 5-hydroxy-2,2,4,6-tetramethyl-3-oxooctanoic acid (HTMOO) residue. The amino acid units and HTMOO residue of **1** were then connected through the analysis of HMBC correlations (Figure 2). The connection between alanine and N-methylglycine was demonstrated by correlations from H-4 ( $\delta_{\text{H}}$  2.92, 4.96) with C-2. The N-methyl ( $\delta_{\text{H}}$  3.16) was correlated with C-5 ( $\delta_{\text{C}}$  170.0) of the HTMOO unit, and the H-9 ( $\delta_{\text{H}}$  5.19) of the HTMOO unit was correlated with C-1 ( $\delta_{\text{C}}$  169.5) in the HMBC spectrum. Therefore, the linkage of alanine and N-methylglycine to the HTMOO was established to satisfy the unsaturation requirement and create a 12-membered cyclic lipodepsipeptide. Accordingly, the whole planar structure of metabolite **1** was elucidated.

The stereochemistry of metabolite **1** was determined by analysis of NOESY data and Marfey's method.<sup>6</sup> The NOESY correlations of H-9/H<sub>3</sub>-18, H-10/H-8, and H<sub>3</sub>-13/H<sub>3</sub>-14 indicated that these protons were cofacial (Figure 2). A detailed analysis of the structure of amino acid units of **1** clearly suggested that only the alanine has chirality. Thus, the stereochemistry of alanine was partly determined by acid hydrolysis, derivatization with Marfey's reagent, and then HPLC analysis. Finally, the configuration of alanine was identified as L-alanine.

A single crystal of metabolite **1** was obtained from acetone after recrystallization. The structure and relative configuration of metabolite **1** was further confirmed by single-crystal X-ray

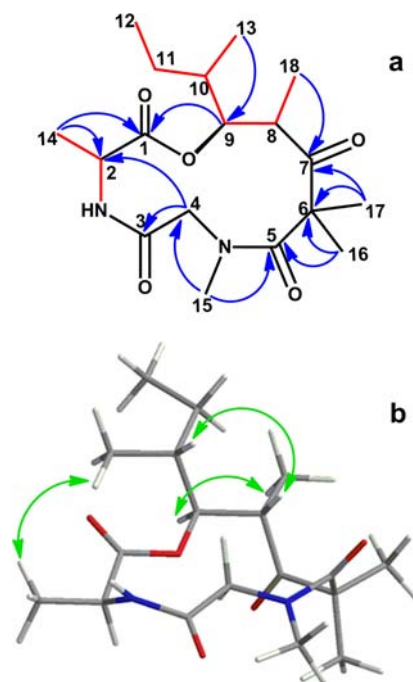


Figure 2. Key (a)  $^1\text{H}$ – $^1\text{H}$  COSY (—), HMBC (H  $\rightarrow$  C), and (b) NOESY correlations of **1**.

diffraction analysis with Cu K $\alpha$  radiation (Figure 3). The  $-0.1$  value of the Flack parameter permitted the establishment of the

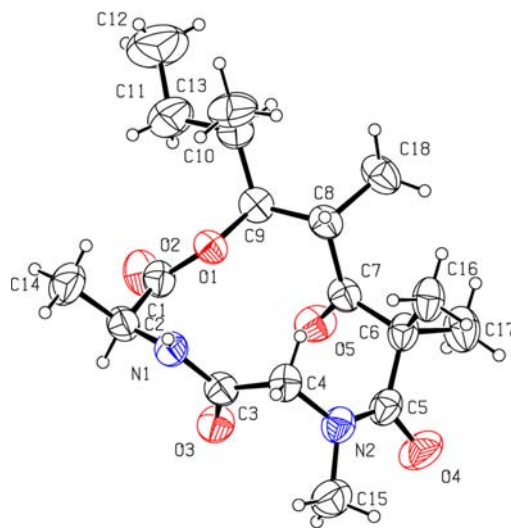


Figure 3. X-ray crystallographic structure of **1**.

absolute configuration of metabolite **1**.<sup>7,8</sup> Therefore, the overall stereochemistry of **1** was assigned as 3*S*,5*S*,6*S*,12*R* and named dahurelmusin A.

Comprehensive spectroscopic analysis revealed that **1** contains the peptide portion and the HTMOO unit, which were derived from nonribosomal peptide synthetase pathways and polyketide synthase origin, respectively.<sup>9</sup> Metabolite **1** incorporated a unique HTMOO unit; no metabolites have been reported to possess this structural feature. Additionally, this was the first time that metabolite **1**, as a 12-membered cyclic hybrid peptide–polyketide, was obtained from the extract of a grass infected by *Epichloë* spp. endophyte. Microtermolide A, a cyclic hybrid peptide–polyketide, is derived from

microorganisms; it was isolated from the fungus *Streptomyces* sp.<sup>10</sup> Accordingly, dahurelmusin A (**1**) is very likely derived from *E. bromicola* living within *E. dahuricus*, and the obtained metabolite was first isolated from the *Elymus* genus.

The insecticidal activity of compound **1** was then assayed against *R. padi* because *R. padi* is one kind of phytophageous on *Elymus dahuricus*. The LC<sub>50</sub> value of compound **1** against *R. padi* was 0.251 ± 0.006 mM. The result suggested that compound **1** displayed obvious insecticidal effects against *R. padi*. To further investigate the possibility of utilizing compound **1** as a broad-spectrum insecticide, we tested its insecticidal activity against *B. brassicae* with an LC<sub>50</sub> value of 0.092 ± 0.007 mM.<sup>11</sup> The results showed that compound **1** has the potential to be exploited as a natural broad-spectrum insecticidal agent resistant to aphids. This is the first report that hybrid peptide–polyketides from *E. bromicola* infected *E. dahuricus* exhibit insecticidal activities against *R. padi* and *B. brassicae*. Therefore, compound **1** could be considered as a lead compound for the development of a new kind of industrial insecticide to suppress aphids. These results will provide a foundation for researchers to find better insecticides from the *Epichloë* spp. infected forage.

## ■ ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b03568.

Experimental procedures; crystallographic data of **1**; 1D and 2D NMR, UV, CD, IR, and HRESIMS of dahurelmusin A (**1**) (PDF)

X-ray crystallographic data of compound **1** (CIF)

## ■ AUTHOR INFORMATION

### Corresponding Authors

\*E-mail: zhibiao@lzu.edu.cn.

\*E-mail: npchem@lzu.edu.cn.

### ORCID

Qiu-Yan Song: 0000-0003-1469-9353

Kun Gao: 0000-0002-3856-3672

### Notes

The authors declare no competing financial interest.

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